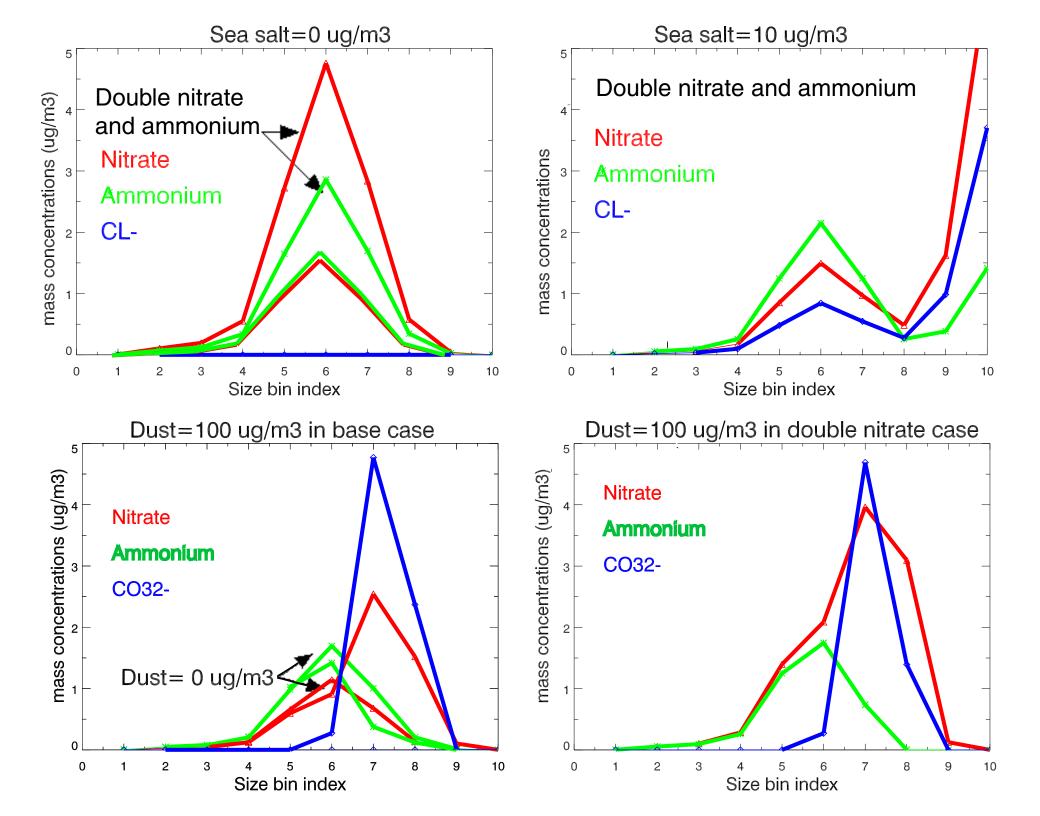
# Nitrate and Ammonium in Aerosols: Effects of Dust and Sea Salt

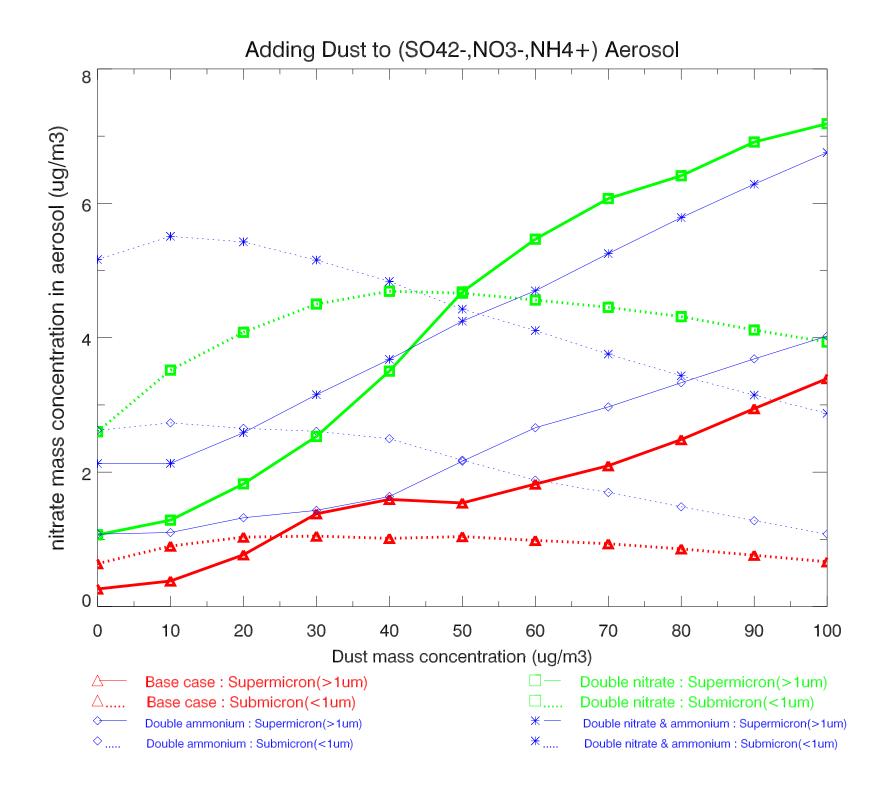
Joyce E. Penner and Y. Feng University of Michigan

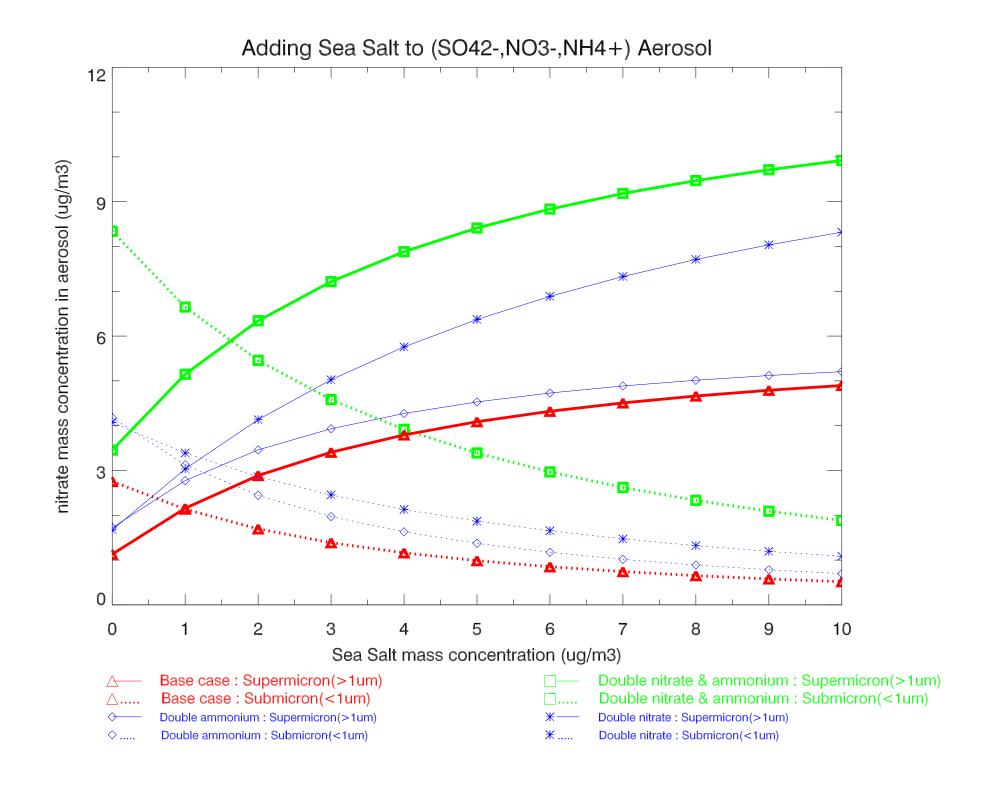
DOE Atmospheric Chemistry Program February 13-15, 2001

#### Motivation.

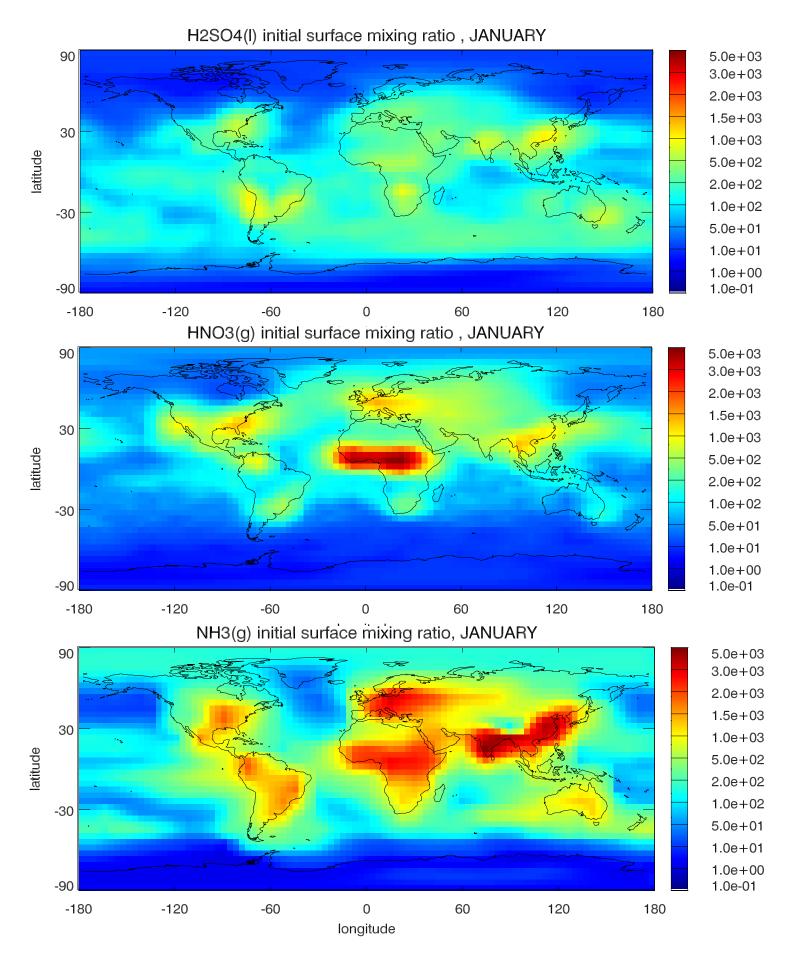
- Nitrate and Ammonium are incorporated in aerosol as a result of chemical interactions with other components of the aerosol
- Inclusion of this equilibrium may lead to changes in aerosol forcing of climate especially in future scenarios where NO<sub>x</sub> emissions may grow.
- The incorporation of nitrate in aerosol may lead to decreases in  $NO_x$ , thereby decreasing formation rates of ozone



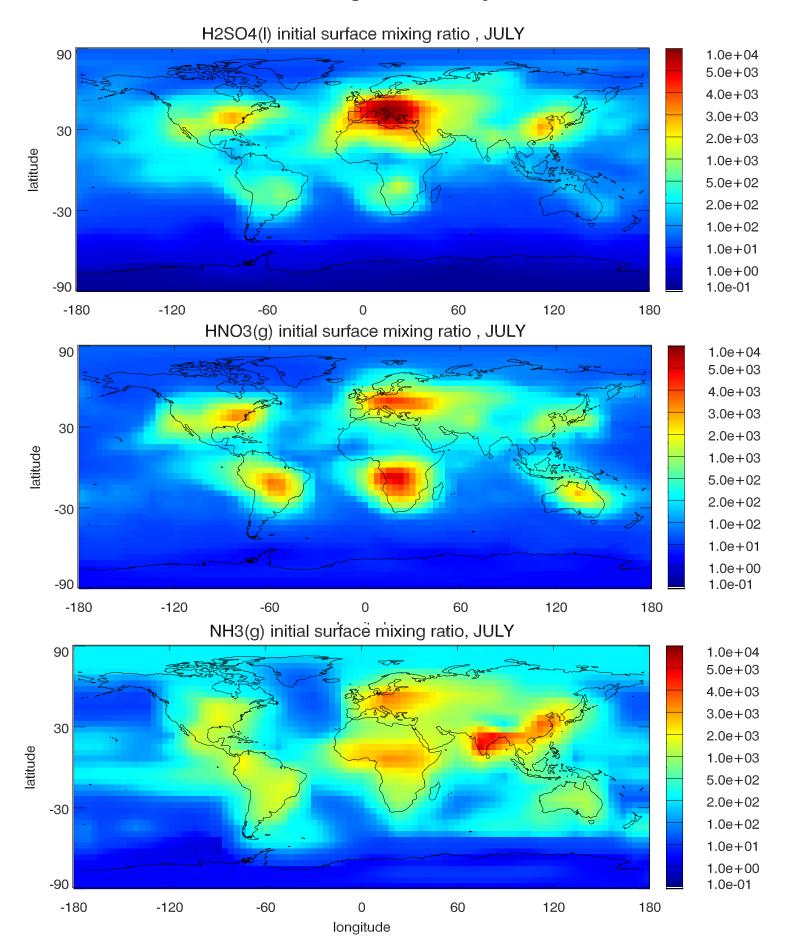


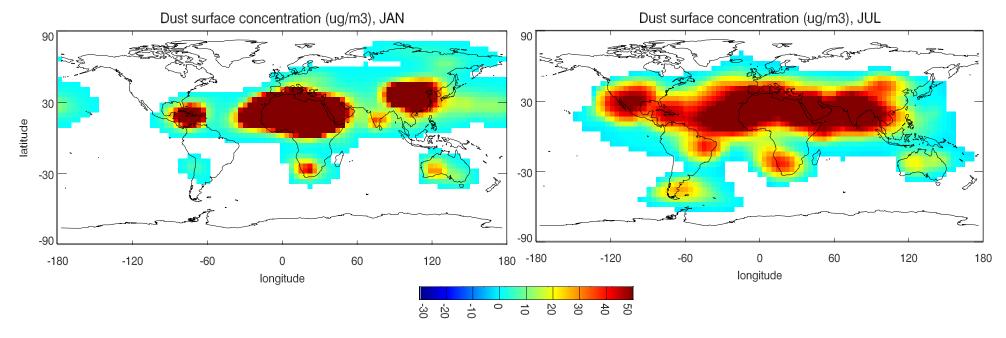


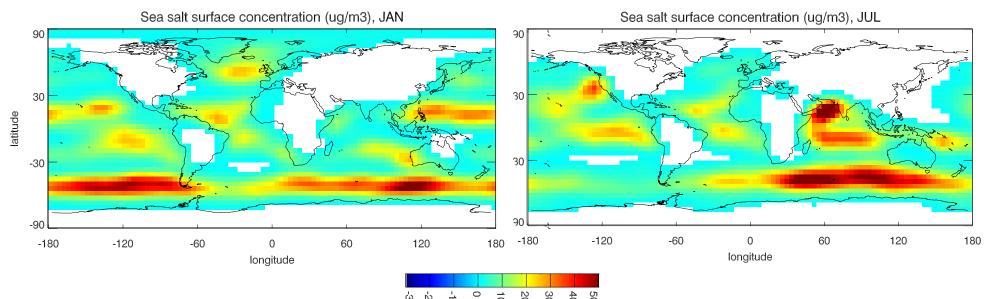
# Initial mixing ratios, January



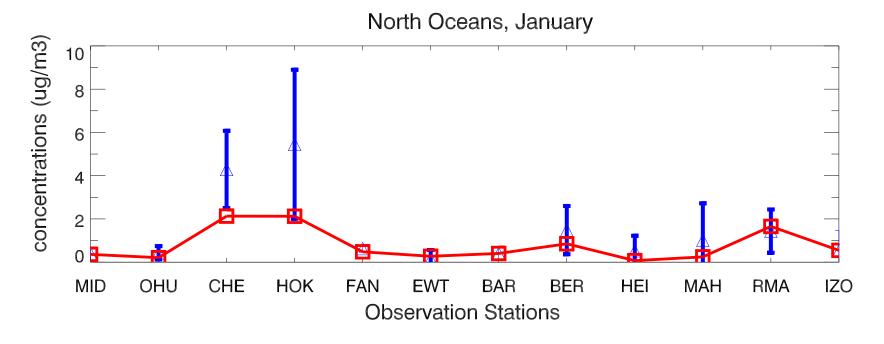
# Initial mixing ratios, July

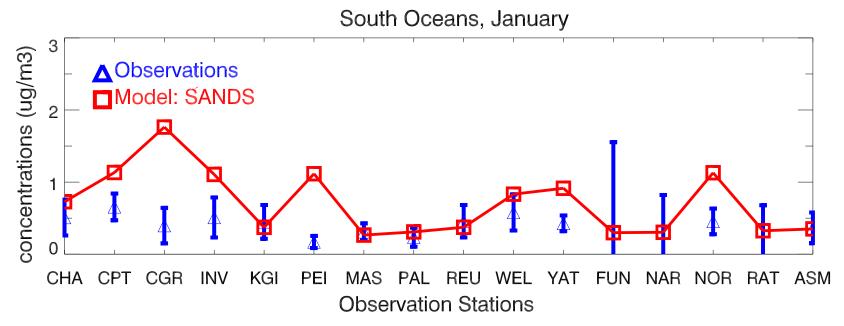




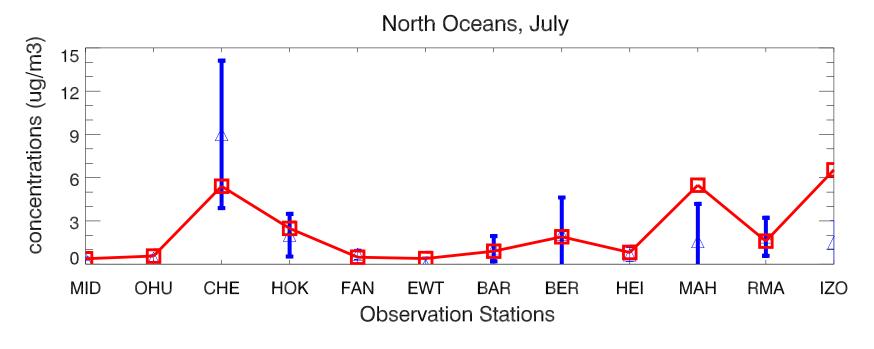


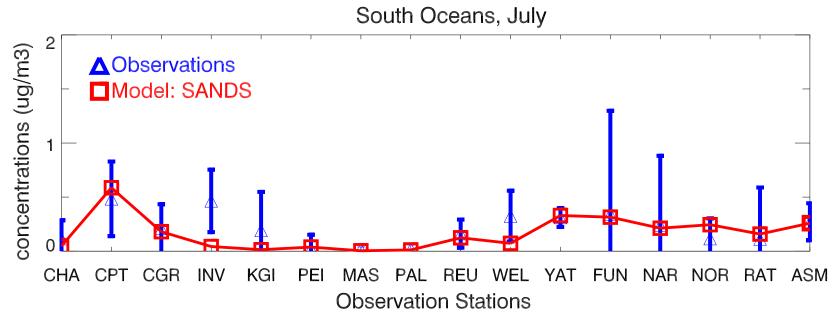
# Comparison of SO4- surface concentrations to observations





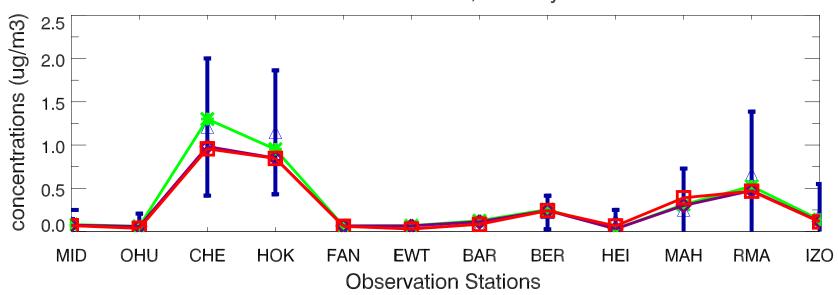
### Comparison of SO4- surface concentrations to observations



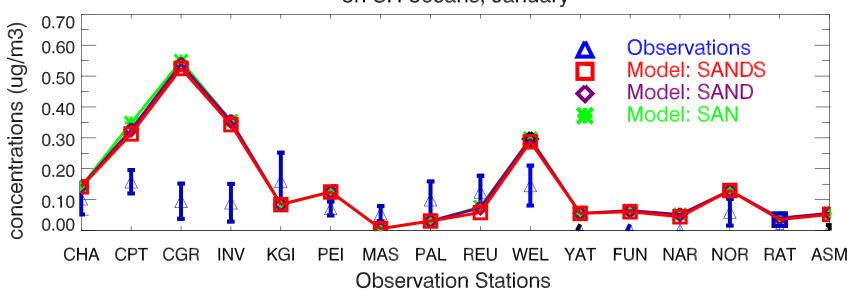


# Comparison of NH4- surface concentrations to observations

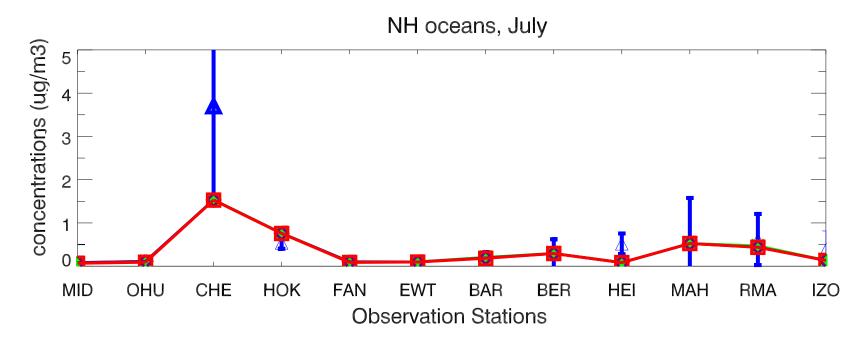


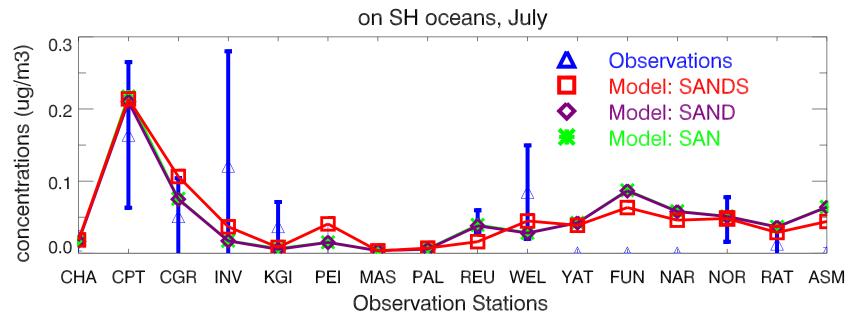




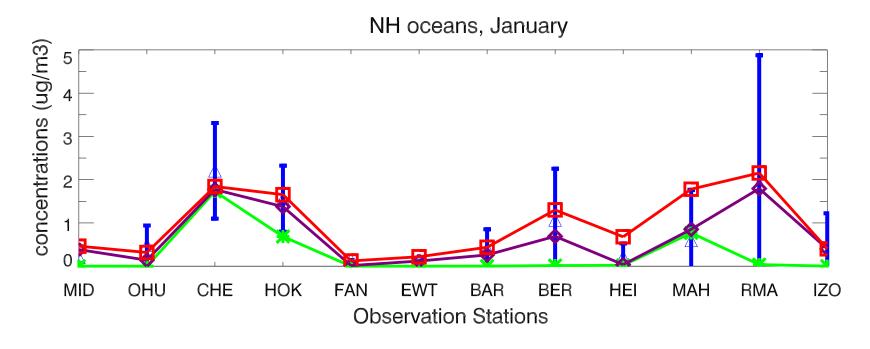


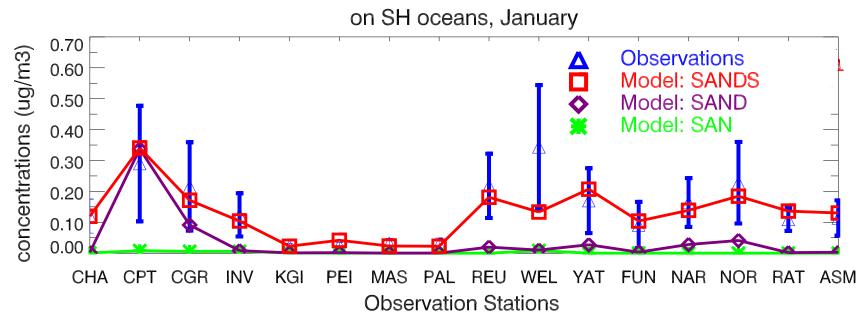
# Comparison of NH4- surface concentrations to observations



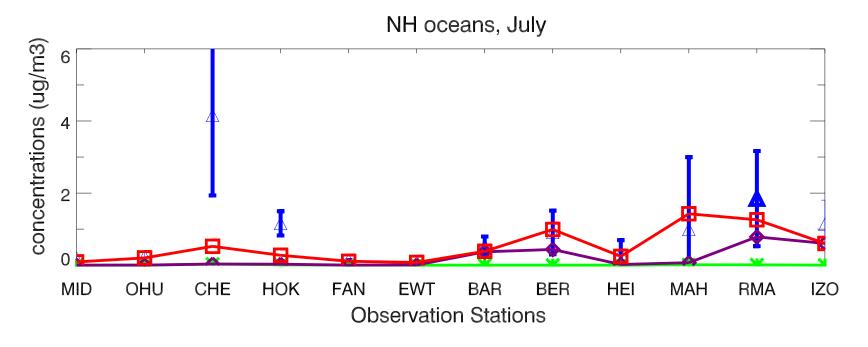


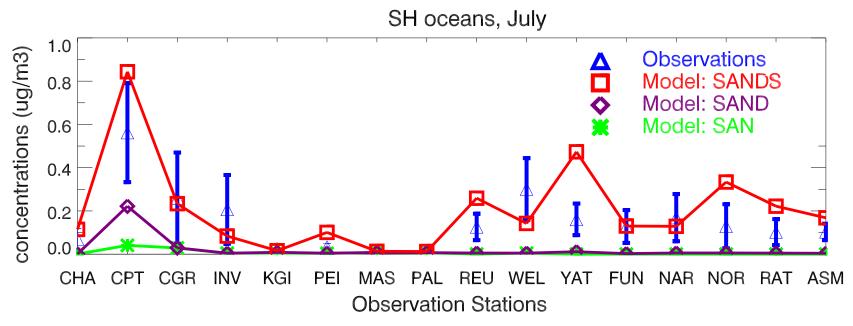
### Comparison of NO3- surface concentrations to observations



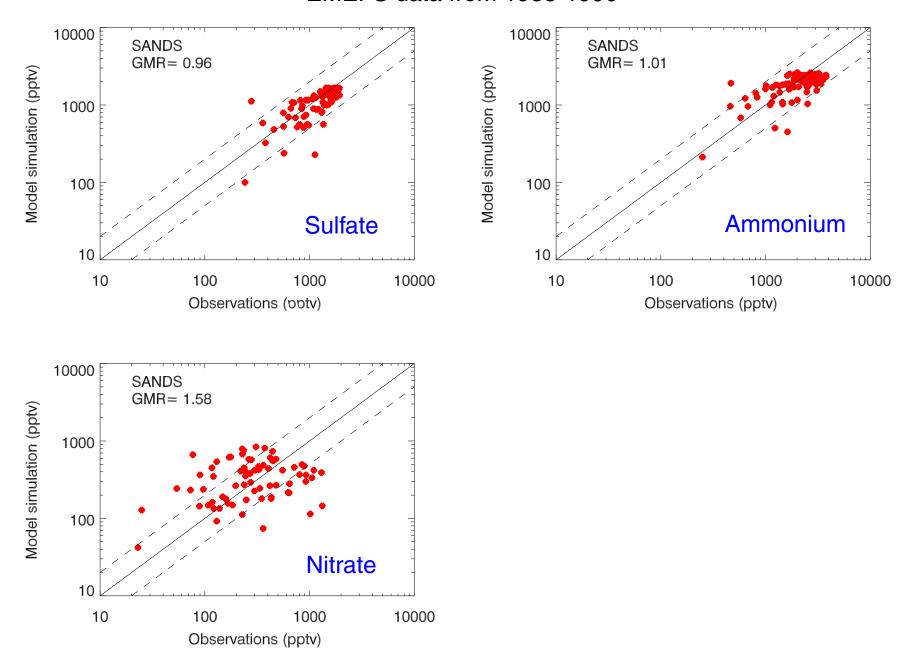


### Comparison of NO3- surface concentrations to observations

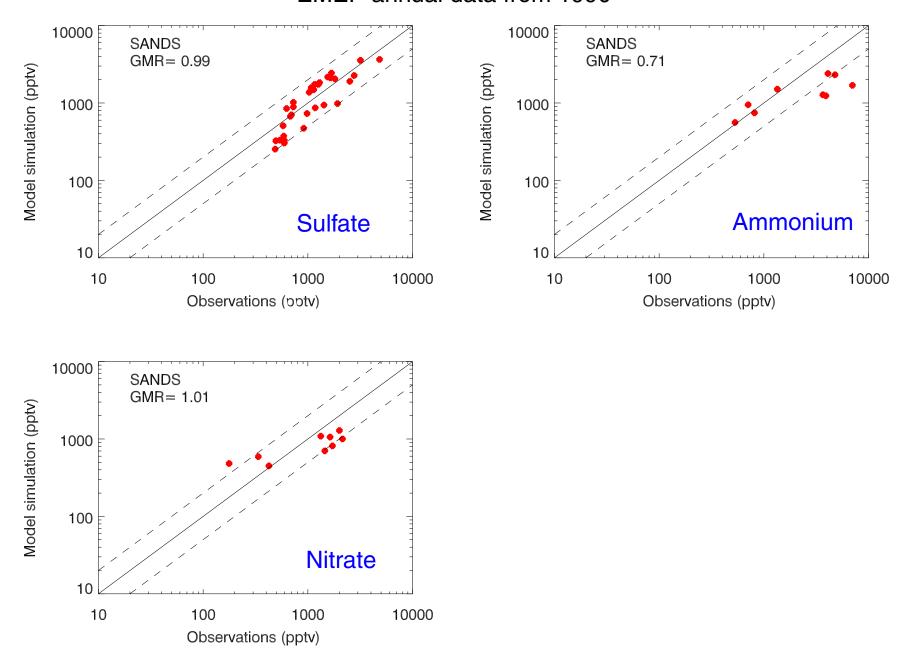




# Comparison of modeled and observed surface concentrations in North America EMEFS data from 1988-1990

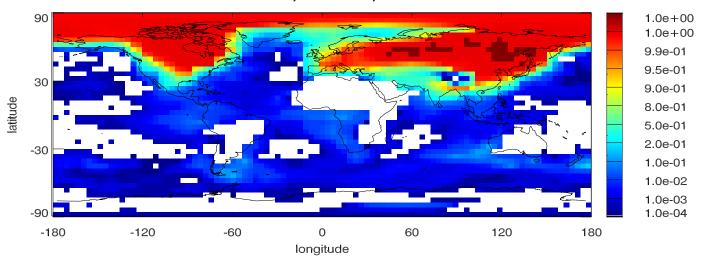


# Comparison of modeled and observed surface concentrations in Europe EMEP annual data from 1990

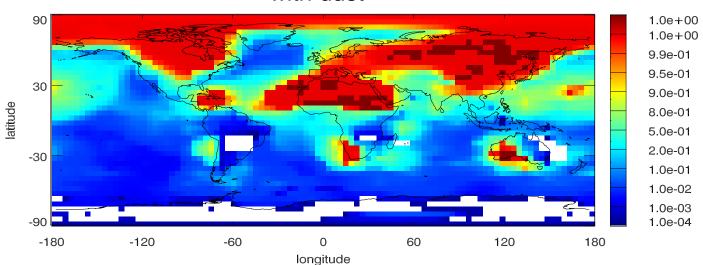


### Fraction of nitrate in aerosol at the surface in January

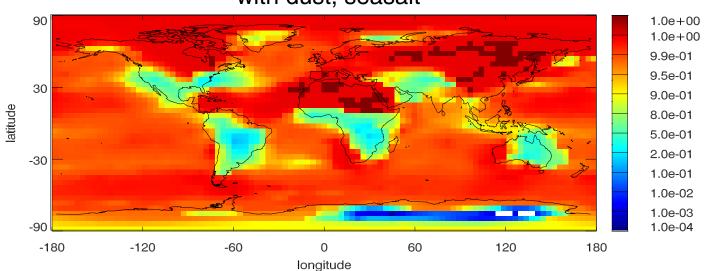




#### with dust

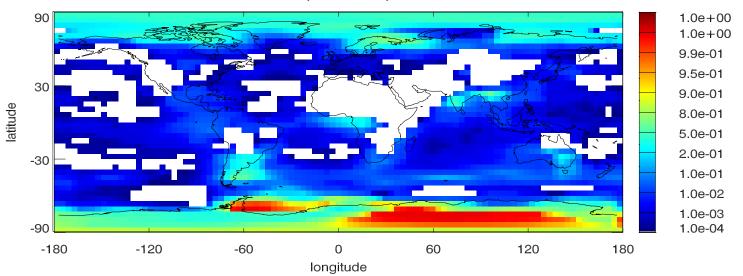


# with dust, seasalt

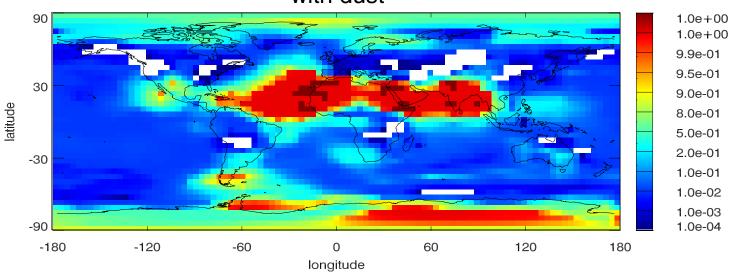


# Fraction of nitrate in aerosol at the surface in July

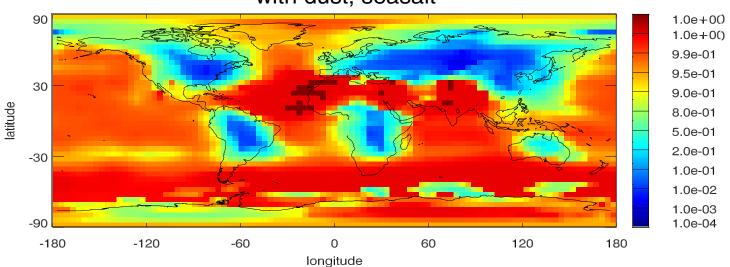




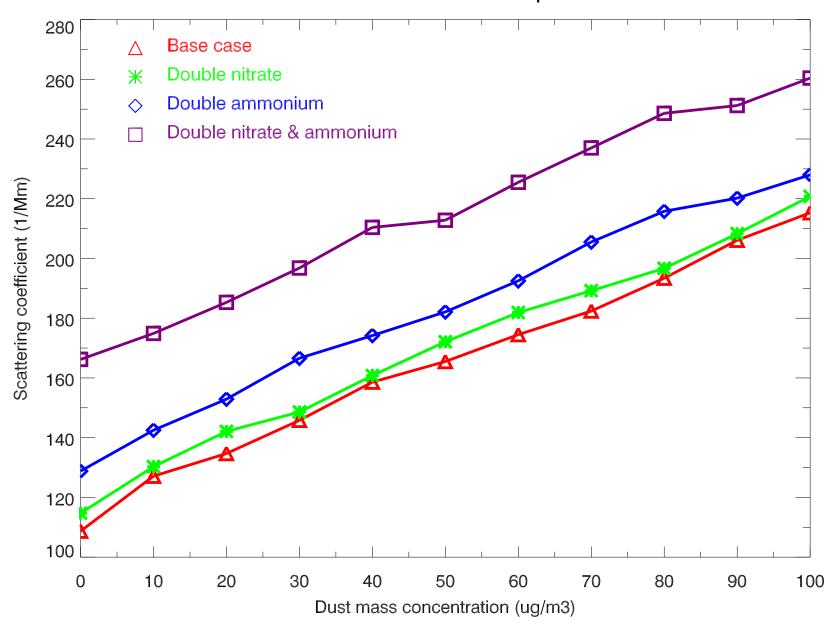
#### with dust



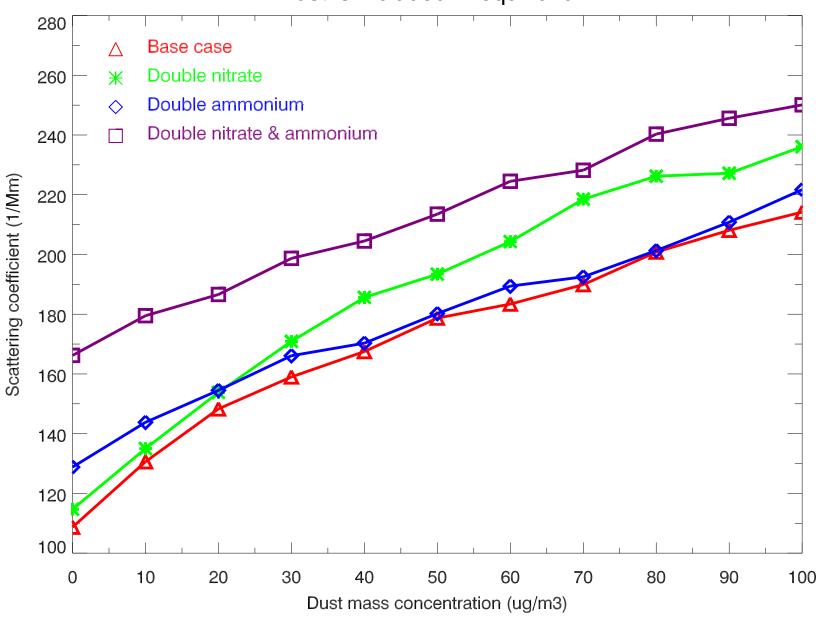
#### with dust, seasalt

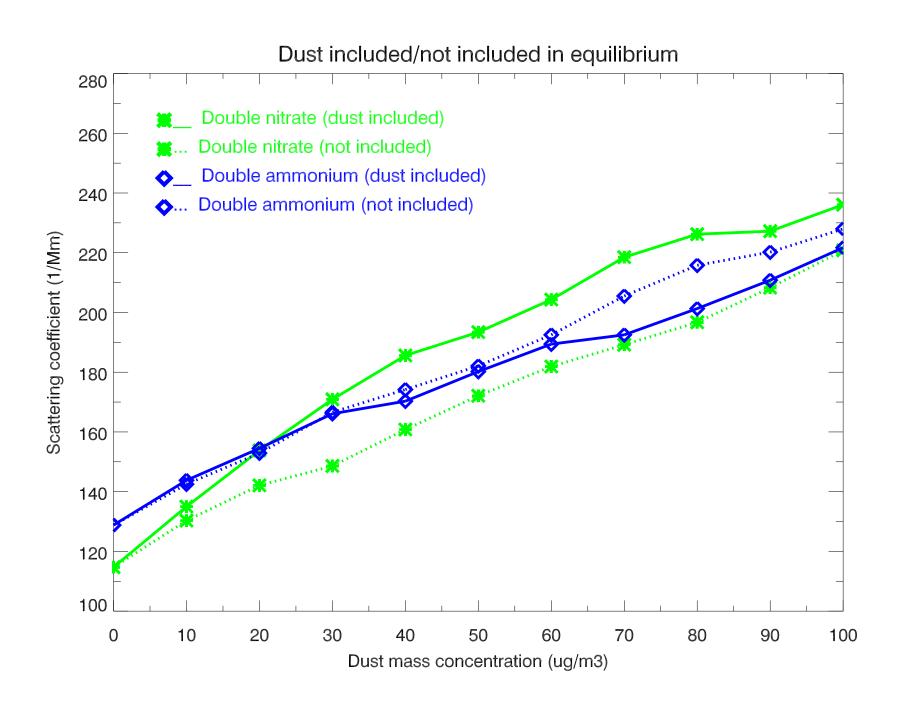


# Dust is not included in equilibrium

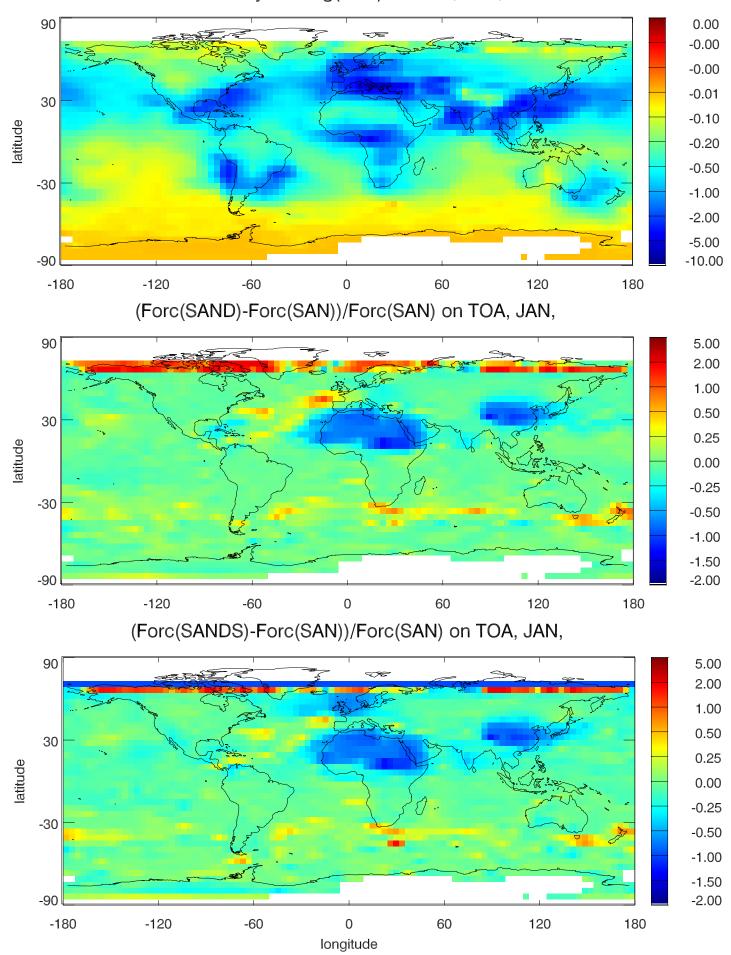


# Dust is included in equilibrium





#### Clear-sky Forcing(SAN) on TOA, JAN,



#### Clear-sky Forcing(SANDS) on TOA, JULY, 90 0.00 -0.00 -0.00 -0.01 30 -1.00 latitude -2.00 -5.00 -30 -10.00 -20.00 -30.00 -90 -40.00 -120 -180 -60 0 60 120 180 (Forc(SAND)-Forc(SAN))/Forc(SAN) on TOA, JULY, 90 5.00 2.00 1.00 0.50 30 0.25 latitude 0.00 -0.25 -30 -0.50 -1.00 -1.50 -90 -2.00 -180 -120 -60 0 60 120 180 (Forc(SANDS)-Forc(SAN))/Forc(SAN) on TOA, JULY 90 5.00 2.00 1.00 0.50 30 0.25 latitude 0.00 -0.25 -30 -0.50 -1.00 -1.50 -90 -2.00 -180 -120 -60 0 60 120 180

longitude

Table: Total aerosol and gas phase burdens in the atmosphere

	H2SO4	NO3-	HNO3	NH4+	NH3	Dust	SeaSalt
	(Tg S)	(Tg N)	(Tg N)	(Tg N)	(Tg N)	(Tg)	(Tg)
H2SO4+HNO3+NH3	0.62	0.12	1.34	0.28	0.38	-	-
H2SO4+HNO3+NH3 +DUST	0.62	0.16	1.34	0.26	0.38	28.83	-
H2SO4+HNO3+NH3 +DUST+SEASALT	0.62	0.25	1.34	0.26	0.38	28.83	5.15

# Estimated forcing associated with different chemical representation of the aerosol (Wm<sup>-2</sup>)

	Forcing based on monthly-averaged-RH	Forcing based on constructed max-RH	Forcing based on daily varying RH
H2SO4	-0.45	-0.56	-0.47
H2SO4+NH3	-0.39	-0.51	-0.47
H2SO4 +NH3+HNO3	-0.42	-0.58	-0.47
H2SO4 +NH3+HNO3 +DUST	-0.40	-0.52	-0.43
H2SO4 +NH3+HNO3 +DUST+SEA SALT	-0.38	-0.50	-0.42
H2SO4 +NH3+HNO3 (Year: 2100)	-0.57	0.83	-0.62

#### Conclusion.

- A significant fraction of the total nitrate and ammonia are incorporated within the aerosol, especially in areas with high dust and sea salt concentrations
- This leads to an overall decrease in climate forcing compared to that calculated for sulfate aerosols alone (-0.42 Wm<sup>-2</sup> compared to -0.47 Wm<sup>-2</sup>)
- If future concentrations of  $NO_x$  emissions continue to increase, the forcing by the combined sulfate, ammonium, and nitrate aerosol may increase to  $-0.62~\rm Wm^{-2}$